

Degenerate Perturbation Theory -

Consider a case like the H-atom for which the eigenenergies are degenerate (n^2 for H-atom e.g. $2s + 3s, 2p$) If we calculate 2nd order perturbed energy & the perturbing potential V "connect" states that are degenerate (e.g. $\langle 2s | V | 2p \rangle \neq 0$) then since $E_{2s} = E_{2p}$ we have a zero in the denominator. The problem essentially is that even for a very small the wavefunction makes a finite shift - i.e. $\partial_{\lambda} \Psi$ does not exist.

Analogy - If we have a flat surface it doesn't matter which directions we take for $x \pm y$. However the Smallest inequality makes particular directions for $x \pm y$ preferred. In some sense if we had initially selected $x \pm y$ congruent with the not-yet-applied inequality there would be no discontinuous change in $x \pm y$ directions when that inequality is applied.

→ We need the "right" set of degenerate eigenfunctions that will be congruent with the applied perturbation. Then there will be no discontinuous change.

The solution is easy. Form the matrix: $\langle i | V | j \rangle$ where $|ij\rangle$ run over the degenerate eigenfunctions ("subspace"). Find the eigenvectors/values of this matrix. The eigenvalues will be E_{ij} ; the eigenvectors (call them $|d\rangle$) will automatically NOT "connect" thru V i.e. $\langle d | V | b \rangle \propto \delta_{ab}$

Note: things can still go wrong if the matrix $\langle i | V | j \rangle$ has degenerate eigenvalues - the eigenvectors $|d\rangle$ will not be uniquely defined. In some higher order we may discover the "right" combination

$$\text{consider: } [v][a] = E_a [a] \quad \& \quad [v][b] = E_b [b]$$

we know $[b]^+ [c] = 0$ since $E_a \neq E_b$ 2nd order safe

$$\text{consider } |a\rangle = \sum a_i |i\rangle \quad |b\rangle = \sum b_i |i\rangle$$

$$\langle b|v|a\rangle = \sum_j \sum_i b_j^* \underbrace{\langle j|v|i\rangle}_{E_a a_j} = E_a [b]^+ [c] = 0$$

similarly if $[a]$ is normalized $\langle 2|v|1\rangle = E_a \leftarrow \text{first order shift}$

Eg: 2d particle in box with $V = \lambda \delta(x - \frac{L}{3}) \delta(y - \frac{L}{4})$

$$E = \frac{\hbar^2 \pi^2}{2m L^2} (n_x^2 + n_y^2) \quad |n_x n_y\rangle = \frac{1}{L} \sin\left(\frac{n_x \pi x}{L}\right) \sin\left(\frac{n_y \pi y}{L}\right)$$

$|12\rangle \& |21\rangle$ are degenerate.

$$\langle 12|v|12\rangle = \lambda \left(\frac{2}{3}\right)^2 \underbrace{\sin^2\left(\frac{\pi}{3}\right)}_{3/4} \underbrace{\sin^2\left(\frac{2\pi}{4}\right)}_1 = \lambda \left(\frac{2}{3}\right)^2 \frac{3}{4}$$

$$\langle 21|v|21\rangle = \lambda \left(\frac{2}{3}\right)^2 \underbrace{\sin^2\left(\frac{2\pi}{3}\right)}_{3/4} \underbrace{\sin^2\left(\frac{\pi}{4}\right)}_1 = \lambda \left(\frac{2}{3}\right)^2 \frac{3}{8}$$

$$\langle 12|v|21\rangle = \lambda \left(\frac{2}{3}\right)^2 \underbrace{\sin\left(\frac{\pi}{3}\right) \sin\left(\frac{2\pi}{3}\right)}_{3/4} \underbrace{\sin\left(\frac{2\pi}{4}\right) \sin\left(\frac{\pi}{4}\right)}_1 = \lambda \left(\frac{2}{3}\right)^2 \frac{3}{4} \frac{1}{16}$$

$$[v] = \lambda \left(\frac{2}{3}\right)^2 \frac{3}{4} \begin{bmatrix} 1 & \frac{1}{16} \\ \frac{1}{16} & 1 \end{bmatrix} \rightarrow \det \begin{bmatrix} 1-\gamma & \frac{1}{16} \\ \frac{1}{16} & 1-\gamma \end{bmatrix} = (1-\gamma)(1-\gamma) - \frac{1}{16} =$$

$$[\epsilon] = \begin{bmatrix} 1 \\ -\frac{1}{16} \end{bmatrix} \quad \text{take } b \perp: \begin{bmatrix} r_2 \\ 1 \end{bmatrix} = \gamma (x - \frac{3}{2}) \rightarrow \begin{matrix} 0 & \frac{3}{2} \\ a & b \end{matrix}$$

$$|\alpha\rangle = \frac{1}{\sqrt{2}} \left\{ \sin\left(\frac{\pi x}{3}\right) \sin\left(\frac{2\pi y}{4}\right) - \sqrt{2} \sin\left(\frac{2\pi x}{3}\right) \sin\left(\frac{\pi y}{4}\right) \right\}$$

\rightarrow has a node @ $(x,y) = (\frac{5}{3}, \frac{5}{4})$

$$|B\rangle = \frac{1}{\sqrt{2}} \left\{ r_2 \sin\left(\frac{\pi x}{3}\right) \sin\left(\frac{2\pi y}{4}\right) + \sin\left(\frac{2\pi x}{3}\right) \sin\left(\frac{\pi y}{4}\right) \right\}$$

\rightarrow is near max @ $(x,y) = (\frac{5}{3}, \frac{5}{4})$

[max @ (.326, .287)]

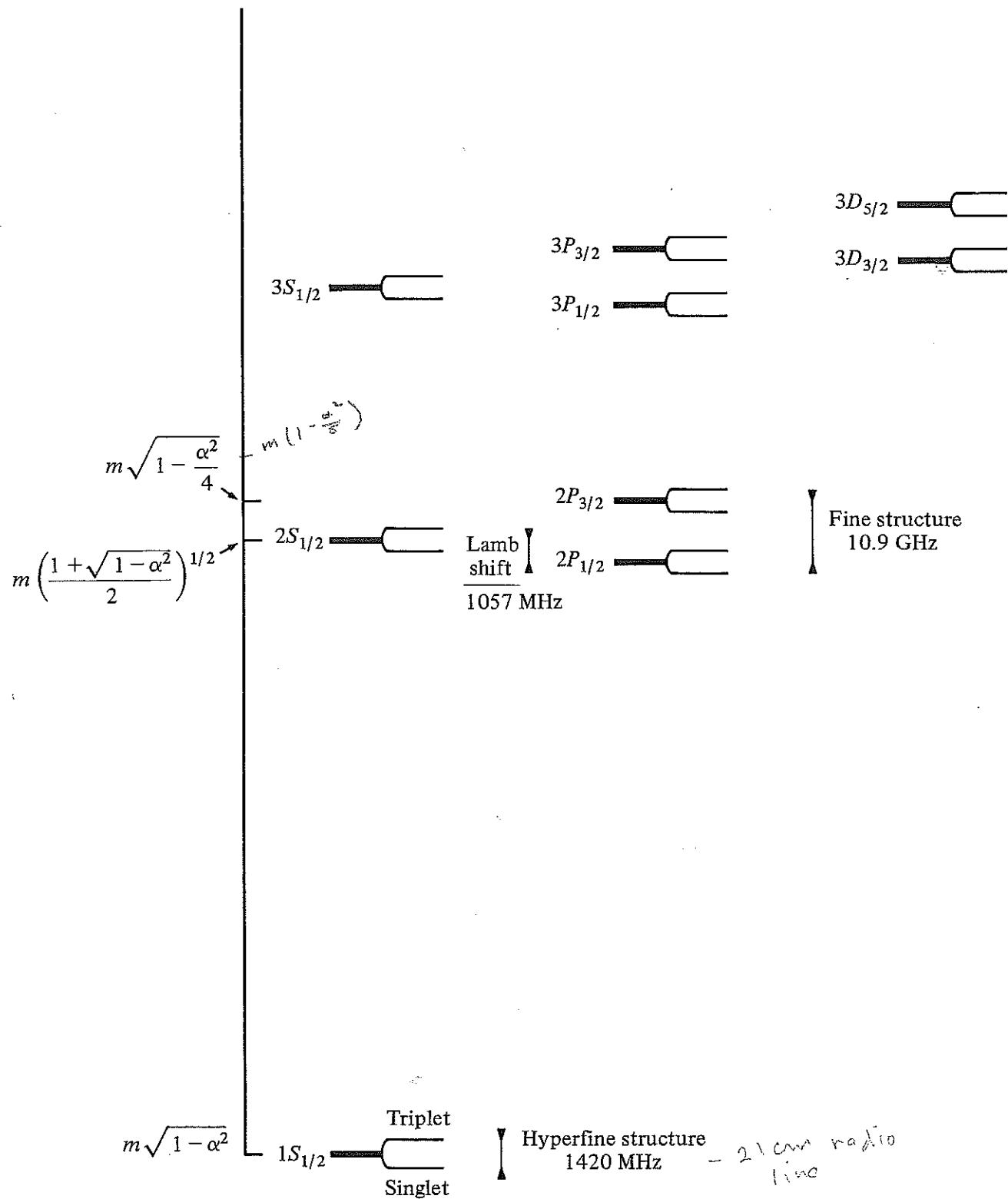


Figure 2-2 Low-lying energy levels of hydrogen.