

Class 38 - Lowry's power plant

$$1) Q = T \Delta S = (342 + 273) (5.3098 - 3.6848) = 10^3 \text{ kJ}$$

$$2) Q = \Delta U - W = \Delta U + P \Delta V$$

$$= (3209 - 2456) + 15 \text{ M}^3 \cdot \text{K} \left(\begin{array}{l} \leftarrow \text{for kJ int} \\ .02491 - .01034 \end{array} \right)$$

$$= 753 + 218 = 971 \text{ kJ}$$

$$= \Delta U + P \Delta V$$

$$= (3297 - 2782) + 1 \text{ M}^3 \cdot \text{K} (.4011 - .2042)$$

$$= 515 + 197 = 712 \text{ kJ}$$

~~3)~~

$$3) Q = \Delta U + \left(\frac{P_1 + P_2}{2} \right) \Delta V = (1586 - \overset{226}{344}) + 7.5 \text{ M}^3 \cdot \text{K} \left(\begin{array}{l} .00166 \\ - .00101 \end{array} \right)$$

$$= 1330 + 5 = 1335 \leftarrow \text{largest } Q$$

$$1365$$

total Q = 4048 kJ

$$4) |W| = |\Delta U| = \begin{array}{r} 3209 - 2782 = 427 \\ 2578 - 3297 = -718 \\ \hline 1145 \end{array}$$

$$5) \frac{W}{Q} = \frac{1145}{4048} = 28\%$$

Note: System operates at high temps between 600 & 342
 not really q
 high temp

$$1 - \frac{273 + 54}{273 + 600} = 62\%$$